

II Semester B.Sc. Examination, June 2008
(Semester Scheme)
PHYSICS – II
Properties of Matter, Heat and Thermodynamics

Time : 3 Hours

Max. Marks : 60

PART – A

I. Answer **any five** of the following. Each question carries **six** Marks. (5×6=30)

1. a) What is internal bending moment ?
b) Deduce an expression for the depression at the loaded end of a single cantilever. (1+5)
2. a) State and explain Bernoulli's theorem.
b) With necessary theory describe stoke's method of determining coefficient of viscosity of a highly viscus liquid. (2+4)
3. a) Define surface tension and surface energy.
b) Obtain an expression for the pressure difference across a curved liquid surface. (2+4)
4. a) State the principle of equipartition of energy.
b) Deduce an expression for the diffusion coefficient of a gas on the basis of kinetic theory of gases. (1+5)
5. a) Distinguish between isothermal and adiabatic processes.
b) Obtain the relation connecting pressure and volume of a gas undergoing adiabatic process. (2+4)
6. a) What is entropy ?
b) Obtain an expression for the change in entropy when two gases are mixed. (1+5)



7. a) Define Helmholtz and Gibb's free energy.
- b) Using Maxwell's thermodynamic relation obtain an expression for the difference in specific heats for a Vander Wall's gas. (2+4)
8. a) Describe porons plug experiment.
- b) Prove that enthalpy remains constant during Joule-Thomson effect. (3+3)

PART – B

II. Answer **any four** of the following : (4×5=20)

- 9) A disc of mass 1 kg and radius 0.1 m is suspended by a wire of length 0.6 m and radius 5×10^{-4} m. If the period of torsional oscillations is 3.9 S. Calculate the rigidity modules of the material of the wire.
- 10) Water is conveyed through a horizontal tube of diameter 0.08 m and length 4 km at the rate of 20 liters per second. Assuming only viscous resistance, calculate the pressure difference required to maintain the flow. η of water = 10^{-3} NS m^{-2} .
- 11) Calculate the height to which water rises in a capillary tube of radius 5×10^{-4} m when it is dipped in a beaker containing water. Surface tension of water is 0.07 Nm^{-1} and the angle of contact is zero.
- 12) Calculate the dipression in the meeting point of ice for an increase of pressure of 2 atmosphere. Specific volume of ice and water at 0° C are $1.091 \times 10^{-3} \text{ m}^3 \text{ kg}^{-1}$ and $1 \times 10^{-3} \text{ m}^3 \text{ kg}^{-1}$ respectively. Latent heat of ice = $338 \times 10^3 \text{ Jkg}^{-1}$
1 atmosphere = 10^5 Nm^{-2} .
- 13) One litre of hydrogen at 27° C and pressure 10^5 Nm^{-2} expands isothermally until its volume is doubled. Find the final pressure and the amount of work done.
- 14) A reversible heat engine of efficiency 40% has its efficiency increased to 50% when the temperature of the sink is lowered by 50K. Find the temperature of the source and the sink.



PART - C

III. Answer **any five** of the following :

(5×2=10)

- 15) a) Explain why steel girders are made in the form of I section.
 - b) Why does a small air bubble rise slowly through a liquid where as the bigger one rises rapidly ?
 - c) Why it is difficult to separate two plates of glass with a thin layer of water between them ?
 - d) Explain, why C_p is greater than C_v ?
 - e) What are the factors on which the cooling due to adiabatic demagnetisation depend ?
 - f) The melting point of ice decreases and that of wax increases with an increase in pressure. Explain.
 - g) Hydrogen gas is subjected to Joule-Kelvin effect at Lab temperature. Will it result in cooling or heating of the gas ? Explain.
 - h) There is always heating during adiabatic compression of a gas. Explain.
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